

# Experimental determination of A<sub>tot</sub> and pK<sub>a</sub> of whole blood of healthy volunteers, patients with sepsis and post-operative patients: an *in vitro* study



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## INTRODUCTION

During acute respiratory acid-base perturbations, pH changes are limited by the non-carbonic buffers, mainly consisting of proteins and phosphates in plasma, plus hemoglobin in whole blood. According to Stewart's physicochemical approach, the total weak acid concentration ( $A_{tot}$ ) is one of the three independent variables determining pH [1]. The amount of dissociated  $A_{tot}$  ( $A^{-}$ ) depends on their acidic dissociation constant ( $K_a$ ), also known in its logarithmic form:  $pK_a = -\log_{10} K_a$ . Experimental estimates for  $A_{tot}$  and  $K_a$  were previously obtained for human plasma of healthy volunteers (17.2 ± 3.5 mmol/L and 0.80 ± 0.60 × 10<sup>-7</sup>,  $pK_a = 7.10$ , respectively) [2]. Of note, while in clinical practice acid-base measurements are performed on whole blood, no data regarding  $A_{tot}$  and  $K_a$  of blood are currently available.

## <u>Objective</u>

To compute  $pK_a$  and  $A_{tot}$  for whole blood of healthy subjects and two different populations of ICU patients, *i.e.* patients with sepsis and post-operative patients.

# METHODS

Blood was collected from 30 volunteers, 30 patients with sepsis [3] and 27 post-operative patients (ICU "Vecla", Ospedale Maggiore Policlinico, Milan and 2 ICUs of FNKV University Hospital, Prague). Hemoglobin, albumin, total proteins and phosphates concentrations were measured. Blood was equilibrated with different gas mixtures to obtain 20 experimental points with  $PCO_2$  ranging between 20 and 120 mmHg. For each subject, the variation of Strong Ion Difference (SID) over  $PCO_2$  was modeled, and the normal value of SID at  $PCO_2$  of 40 mmHg (SID<sub>40</sub>) was computed. Measured pH and  $PCO_2$ , SID<sub>40</sub> and the simplified strong ion electroneutrality equation were used to calculate  $A_{tot}$  and  $K_a$  through the Marquardt nonlinear regression procedure:

$$0.0307 \cdot PCO_2 \cdot 10^{pH-6.120} = SID_{40} - \frac{A_{tot} \cdot K_a}{K_a + 10^{-pH}}$$

T-test and Mann-Whitney rank sum test were used for analysis.

#### RESULTS

Age [54 ± 15 vs. 61 ± 16 vs. 57 ± 18 yr, p = 0.3] and gender [14 (47%) vs. 9 (33%) vs. 12 (40%) of females, p = 0.6] did not differ among volunteers, post-operative and septic patients. Both populations of ICU patients had lower hemoglobin, albumin and total proteins concentrations as compared to healthy volunteers (**Table 1**), while phosphates were similar. Septic patients had lower values of  $A_{tot}$  as compared to post-operative patients, which had lower values as compared to healthy controls. Also, pK<sub>a</sub> showed a decreasing trend going from controls to post-operative and septic patients.

	Controls (N=30)	Post-operative patients (N=27)	Septic patients (N=30)	р
Hemoglobin (g/dL)	14.4 ± 1.0	10.9 ± 1.8	9.6 ± 1.3	< 0.00
Albumin (g/dL)	4.8 ± 0.3	$3.2 \pm 0.4$	2.5 ± 0.4	< 0.00
Total protein (g/dL)	7.2 (6.7 – 7.6)	4.8 (4.6 – 5.1)	4.8 (4.1 – 5.1)	< 0.00
A <sub>tot</sub> (mmol/L)	94 (72 – 132)	53 (43 – 62)	39 (37 – 46)	<0.001
рК <sub>а</sub>	7.91 (7.84 – 8.19)	7.74 (7.52 – 7.82)	7.35 (7.11 – 7.46)	<0.001

<u>**Table 1**</u>. Comparison between nonvolatile buffers concentration and the estimated  $A_{tot}$  and  $pK_a$  values among groups.

## **CONCLUSIONS**

Healthy volunteers, septic patients and post-operative patients had different values of both  $A_{tot}$  and  $pK_a$  of whole blood. Of note, both estimates performed through whole blood equilibration were remarkably different as compared to the ones previously obtained on plasma. Interestingly, the values of  $pK_a$  we computed were higher than 7.40, suggesting a higher non-carbonic buffer power of blood against acute respiratory alkalosis rather than acidosis.

### REFERENCES

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